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Research Article

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An *in vitro* Study to Assess the Potential of a Unique Micro porous Algal Derived Cap Bone Void Filler in Comparison with Clinically-Used Bone Void Fillers

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Abstract

Macroporosity(>100µm) in bone void fillers is a known prerequisite for tissue regeneration, but recent literature has highlighted the added benefit of microporosity(0.5 - 10µm). The aim of this study was to compare the *in vitro* performances of a novel interconnective microporous hydroxyapatite (HA) derived from red algae to four clinically available macroporous calcium phosphate (CaP) bone void fillers. The use of algae as a starting material for this novel void filler overcomes the issue of sustainability, which overshadows continued use of scleractinian coral in the production of some commercially available materials, namely Pro-Osteon™ and Bio-Coral®. This study investigated the physicochemical properties of each bone voidfiller material using x-ray diffraction, fourier transform infrared spectroscopy, inductive coupled plasma, and nitrogen gas absorption and mercury porosimetry. Biochemical analysis, XTT, picogreen and alkaline phosphatase assays were used to evaluate the biological performances of the five materials. Results showed that algal HA is non-toxic to human foetal osteoblast (hFOB) cells and supports cell proliferation and differentiation. The preliminary *in vitro* testing of microporous algal-HA suggests that it is comparable to the four clinically approved macroporous bone void fillers tested. The results demonstrate that microporous algal HA has good potential for use *in vivo* and in new tissue engineered strategies for hard tissue repair.

Keywords: Bone Void Filler; Calcium Phosphate Bioceramic; Algae; hFOBS

Introduction

The purpose of bone void fillers, besides filling a bone defect, is to encourage new bone growth by providing structural support for blood clotting and bone remodelling [1]. Ideally, the bone void filler should be resorbed and replaced through sequential remodelling cycles that are synchronised with the natural remodelling process. This will ensure that mechanical stability and metabolic function are maintained at the defect site and minimise the risk of failure from infection [2,3]. Over recent years, an increasing number of synthetic bone substitutes have become clinically available as alternatives to traditional graft materials [4,5]. Despite this, clinicians preferentially choose autologous bone graft, as synthetic alternatives still fail to match their *in vivo* performance [6,7].

The majority of synthetic bone void fillers tend to be HA, or related CaP ceramics, due to their similarity to bone mineral, which is approximately 60% HA by weight [8]. Research into naturally derived and synthetic CaP bone void fillers has focused mainly on the design and regulation of material properties to augment tissue regeneration. In more recent years, the focus has shifted towards the “*intelligent biomaterial*” which delivers active pharmacological compounds and/or biological entities, such as morphogenic proteins, genes and growth factors, to enhance and accelerate regeneration [9].

There is, however, still a lack of understanding of critical physicochemical factors, which influence specific biological responses. For example, the role of microscale (0.5 - 10µm) and macroscale (>100µm) porosity to induce ectopic bone formation [10,11]. It has been well established that macroporosity with pores and interconnections >100µm improves osteogenic outcomes [12]. The role of microporosity and/or multi-scale porosity in osteogenesis is still not clear. Several studies have reported the negative effects of microporosity with regard

to fibrous ingrowth, reduced cell viability and restricted cell ingrowth [13,14]. Other more recent *in vivo* studies, however have clearly found that microporosity enhances bone regeneration by modulating osteogenic differentiation [15,16], enhancing mechanical properties with more efficient load transfer [17], and improving bioresorption [18].

Mineralised biological tissues in nature, such as bone, teeth, mollusk shells, and coral, have shown that hierarchical structures with interconnected meso (2 - 50nm), micro (0.5 - 10µm) and macro (>100µm) pores are crucial to biological function [19]. The ability to replicate these hierarchical structures in purely synthetic biomaterials has proven to be one of the most demanding manufacturing challenges in the field. Several techniques exist to produce porous bioceramics; but techniques to produce interconnected porous structures are much more limited. Templating is the only manufacturing technique that has shown any real potential, however it relies heavily on the existence of suitable templates [20,21] - those currently investigated include polymer foams, hydrogels, emulsions, composites on block copolymer and surfactants, bovine bone, marine sponges and corals [20,22].

Pro-Osteon™ and Biocoral® are two coral-derived commercially available products currently on the market. These materials have been

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used clinically for more than 10 years and have performed similarly to other synthetic bone void fillers, exhibiting limited osteoinductivity *in vivo* [23,24]. The main benefit of coral-derived bone void fillers over synthetic alternatives is their bioresorption capacity [25,26]. Recently, other sustainable calcified marine structures and organisms, such as mussels [27], cuttlefish bone [28] and algae [26,29,30] have been investigated for bone void filler applications.

The aim of this study was to evaluate microporous (0.5 - 10 μ m) marine-based bioceramic granules (QUB HA) in comparison to other more macroporous (>100 μ m) bovine (Bio Oss®), synthetic (Bone Save) and marine-based (Pro-Osteon™) commercial bioceramic granules. Our hypothesis is that the microporous granules will perform just as well as those with macroporosity in terms of the biological response. Each of these products has been manufactured differently [31-33], giving them a unique set of physicochemical attributes (Table 1). In this initial study, the physicochemical properties were studied using a wide range of analytical techniques [x-ray diffraction (XRD), fourier transform infrared spectroscopy (FTIR), inductive coupled plasma-mass spectrometry (ICP-MS), nitrogen gas absorption (S_{BET}), mercury intrusion porosimetry (MIP)] to obtain a comprehensive overview of each material and preliminary *in vitro* testing of cell attachment, cell death, proliferation and differentiation.

Materials and Methods

Materials

Corallina officinalis was harvested from Fanad, Co. Donegal. It was processed into a biphasic calcium phosphate microporous bioceramic (QUB HA) by a low temperature-pressure synthesis technique reported previously [34,35]. Briefly, the algae samples were cleaned using: sodium hypochlorite (NaOCl; 0.1M), hydrogen peroxide (H₂O₂; 30 wt. % in H₂O) and sodium hydroxide (NaOH, 0.5M), to remove macroscopic dirt/impurities; then heat treated to isolate the inorganic compound. Heat treatment resulted in the partial conversion of algae from Mg-rich-calcite to calcium oxide [36]. For synthesis, a stoichiometric ratio (Ca/P=1.67) of heat treated algae was added to a 2 molar

Diammonium hydrogenphosphate (99.9%; Sigma Aldrich) aqueous solution. The pH was adjusted in the range of 10-12 with ammonium hydroxide (28% in H₂O, Sigma Aldrich) and maintained throughout synthesis. The suspension was heated to 100°C in an open system reaction vessel under constant agitation for 12hrs. It was then cooled and aged for 24hrs at room temperature. The residual phosphate was removed by soaking the granules in 10% v/v CH₃COOH for 1 hr under agitation, which effectively removes the residual phosphate without dissolving synthesised phosphate [34,35]. The resultant granules were washed in dH₂O until the wash solution reached a pH ~ 7, filtered and dried overnight at 90°C. Four different clinically available bone void filler materials were evaluated: 1) Bio Oss®, GeistlichPharma; 2) Bone Save®, Stryker Osteonics; 3) Pro-Osteon™ 200R and 4) Pro-Osteon™ 500R, Biomet (Interpore) and compared to QUB HA.

Material characterization

Scanning Electron Microscopy (SEM) / Image Analysis (IA): SEM coupled with IA was used to determine pore size within the bone void filler materials. Granules were dispersed on the surface of an aluminium SEM stub coated with *Araldite* epoxy glue and then sputter coated with gold. Micrograph images were taken on a JEOL 6500 SEM microscope and imported into imaging software (Lucia). Manual point-to-point (n=20) measurements were taken in each direction to determine the pore size, per image with 3 replications.

Nitrogen gas absorption (S_{BET}): The specific surface areas, m² g⁻¹(S_{BET}), of the granules were determined using nitrogen adsorption isotherms measured on a FisonsSorpomatic 1900 analyser. The areas were calculated using Brunauer, Emmett and Teller (BET) multilayer method (Quantachrome™ NovaWin2 software). Before analysis, the granules were outgassed at -73K for 24hrs.

Mercury intrusion porosimetry (MIP): A mercury porosimeter (QuantachromePoremaster) with a small penetrometer was used to determine the porosity of the bone void filler materials. Samples (0.2–0.4gs) were analysed with a contact angle of 140° and surface tension of 480.00 ergcm⁻². The theoretical porosity was calculated from the total intrusion volume.

Inductively coupled plasma-mass spectrometry (ICP-MS)

Elemental analysis was performed to determine Ca, P and Mg concentrations to calculate the Ca/P and Ca+Mg/P molar ratios using a Perkin Elmer Optical Emission Spectrometer, Optima 4300DV. Prior to analysis the machine was calibrated to 10ppm with a detection limit of 0.01mg L⁻¹. Powdered samples were digested in 2%v/v nitric acid (ACS reagent, ISO, ≥69%) in dH₂O overnight and then diluted within the detection limit of the machine for analysis.

Powder X-ray diffraction (XRD): The bulk chemical composition of the materials was determined using XRD. Powder diffraction scans were recorded at room temperature using an X-Pert Pro X-ray diffractometer system with an X'Celerator X-ray detector (PANalyticalX'Pert PRO MPD; Model No. 135 PW3040/60) and CuK α radiation. For phase identification, a step size of 0.02 and count rate of 0.0005 min⁻¹ were used to obtain the spectra for between 10 and 70° (2 θ). Scans were imported into X-Pert high score for analysis with reference data, then characterised using Rietveld refinement analysis.

Fourier Transform Infra-Red spectroscopy (FTIR) : FTIR was used to identify functional groups on the surface of the bone void filler materials. IR spectra were recorded on a Perkin-Elmer 1600 M-80 spectrometer. Spectra of powdered samples were collected in transmission mode from 124 scans, over a frequency range of 400 to 4400cm⁻¹, at a resolution of 4cm⁻¹.

Biological characterization

Cell culture: Cell culture experiments were performed using a transformed human fetal osteoblast (hFOB) cell line (LGC/ATCC, UK). Cells were cultured in DMEM/F-12 medium supplemented with 10% fetal bovine serum (FBS), 0.3mgml⁻¹ geneticin and 200mM L-glutamine (all Invitrogen, UK). 50 μ M ascorbate-2-phosphate (Sigma,UK) was also added to the medium for the alkaline phosphatase (ALP) experiment only. All assays were incubated in 5% CO₂ in air at the permissive (proliferative) temperature of 33°C (unless otherwise stated). All materials were sterilized by autoclaving at 121°C for 30 minutes in their original granular form. For each experiment, 100 to 120mg/well (\approx 80%) of a 96-well plate was evenly covered in bone void filler granules and then pre-wetted in culture medium for 24hrs. Before seeding, the medium was removed and granules were washed twice in phosphate-buffer saline (PBS). In each assay, the granules and tissue culture polystyrene (TCP) without granules (control) were seeded with hFOBs at a density of 1x10⁵ cells/cm² and fed bi-weekly with the appropriate medium. There were six replicates per experimental condition (n=6) and for each assay readings were performed in duplicate for each well.

Cytotoxicity: Cytotoxicity was measured after 72hrs incubation using a CytoTox96 non-radioactive assay (Promega) according to the

manufacturer's instructions. This assay measures lactate dehydrogenase in the culture supernatant. Absorbance was read at a wavelength of 450nm with a reference wavelength of 620 on a GENios plate reader (TECAN, Theale, U.K.)

Cell viability: Cell viability was observed colorimetrically at 1, 3 and 7 days using an XTT assay (*In vitro* Toxicology Assay Kit, TOX2, and Sigma). Viability was determined by the ability of metabolically active cells to reduce tetrazolium salts (2, 3-bis [2-Methoxy-4-nitro-5-sulphophenyl]-2H-tetrazolium-5-carboxyanilide inner salt) to an orange formazan dye. The reagent was prepared according to manufacturer's instructions and a volume equal to 20% of the culture medium was added to each well. After 4 hrs incubation at 33°C, 50µl aliquots were transferred in duplicate to a fresh 96-well plate and absorbance read at a wavelength of 450nm with a reference wavelength of 620nm. Average absorbance values from medium-only wells were subtracted from net absorbance values (620 - 450nm) for each experimental group.

Cell proliferation: Cell proliferation was determined by measuring the DNA content of cell lysates using a PicoGreen® dsDNA Quantification Kit (Invitrogen). At each time point (1, 3 and 7 days) the culture medium was replaced with 0.1%v/v Triton X-100 in PBS and frozen at -80°C. Cells were lysed by three freeze/thaw cycles and the assay performed as per manufacturer's instructions. Samples were analysed at wavelengths of 480nm (excitation) and 520nm (emission).

Cell differentiation: ALP enzymatic activity was used as a maker of cell differentiation. ALP activity was measured using the substrate *p*-nitrophenyl phosphate (Sigma-Aldrich, UK) in an end-point assay. After seeding hFOB onto each material in complete medium supplemented with ascorbate-2-phosphate, cells were incubated at 33°C for 24hrs, followed by incubation at the restrictive (differentiating) temperature of 39°C for a further 3 and 7 days. Cells were lysed as described in section 2.3.4 and 50l aliquots were added in duplicate to a 96-well plate with 200µl of *p*-nitrophenyl phosphate and incubated for 30 mins. The reaction was stopped by the addition of 1M NaOH solution and absorbance read at a wavelength of 405nm. The DNA content in each lysate was also measured using the PicoGreen® assay (described in section 2.3.4) and the ALP activity results are expressed per µg of DNA.

Statistics

Differences between materials and between time points were analysed using one-way ANOVA with post hoc Bonferroni's test using SPSS v13 software. Where individual comparisons were performed, an independent Student's t-test was used. *p* values <0.05 were considered significant. All data are expressed as mean ± standard error.

Results and Discussion

Physicochemical characterization

This study compared the physicochemical properties and *in vitro* biological performance of five different bone void filler materials. The materials tested differ considerably with respect to their granular size, shape, pore size and chemical composition making direct comparison between specific material properties difficult when assessing their biological performance. When using granules as fillers in a bone defect, it is hypothesized that interstitial space between the granules will create additional macropores, enabling infiltration of bone-forming cells around the granules. Therefore, the shape and size of the granules is important, as it will influence the size of the interstitial pores. For penetration and ingrowth of bone cells, pores and pore interconnections greater than 100µm are required [1,12]. Metabolizing cells also require an oxygen source in close proximity for survival; therefore the ability of the bone void filler to become vascularised will enhance the successfulness of the implant [37].

Morphological analysis: Four of the materials are classified as predominately macroporous with pores in the range of 100 - 900µm. Slight variation in pore size was observed between pore size quoted by the manufacturer (Table 1) and those values calculated in-house (Table 2), owing to variation in characterisation techniques. Only QUB HA was classified as microporous, with pores in the range of 9 - 11µm (Table 2). It also had relatively small granules in the range of 0.3 to 2mm, similar to Bio Oss® and Pro-Osteon™200R, however, the granule shape and type of pores differs between commercial materials (Table 1). This is also evident in the SEM micrographs, Figure 1a-e. The morphological structure of Bio Oss® and Pro-Osteon™200R/500R is similar to cancellous bone, whereas Bone Save® has a lower porosity with mainly closed macropores. The granule

Material	Bio Oss®	Bone Save®	Pro Osteon™ 200R	Pro Osteon™ 500R
Supplier	Geistlich Pharma	Stryker, Osteonics	Biomet, (Interpore)	Biomet, (Interpore)
Type	Bovine	Synthetic	Coral	Coral
Composition	Similar to human bone (HA)	80%TCP: 20% HA	HA-coated CaCO ₃	HA-coated CaCO ₃
Porosity	70 – 75%	50 – 55%	50%	65%
Pore Size (µm)	100	300 – 500	180 – 220	270 – 650
Pore Type	Interconnective	Closed Pores	Interconnective	Interconnective
Shape	Irregular	Rounded	Rounded	Irregular
Granule Size (mm)	0.25 – 2	2 – 8	0.5 – 1	2 – 10

Table 1: Physicochemical material properties of commercial CaP bone void fillers according to manufacturers' specifications.

Material	Bio Oss®	Bone Save®	Pro Osteon™ 200R	Pro Osteon™ 500R	QUB HA
QUB HA Granule Size (mm)	0.25 – 2	2 - 8	0.5 - 1	2 – 10	0.3 - 2(φ) ^c
Pore Size (µm) ^b	200 - 900	300 - 500	150 - 220	200 - 650	9 - 11
Surface Area (m ² g ⁻¹) ^c	94.60	2.29	9.13	9.38	129.8
Porosity ^d	90.65±1.97	85.01±5.61	87.17±7.13	88.67±11.84	88.47±3.21

Table 2: Morphological characterization of CaP bone void fillers measured in-house.

- a. b. Calculated by image analysis (n = 20)
c. Calculated using BET (r² = 0.998; n = 3)
d. Calculated by MIP Theoretical Porosity (n = 3)

shape of QUB HA (Figure 1ei) is cylindrical with closely-spaced honeycomb-like pores and typical pore wall thickness of 1.5 - 2 μ m, pore diameter of >12 μ m (Figure 1eii) and pore length of >100 μ m (parallel to branch). These pores are interconnected by microperforations of 1 - 4 μ m, as characterised in a previous study [35]. According to the literature, this type of microporosity is unlikely to support osteogenesis [13]. Micropores in the range of 0.2 - 1 μ m are also visible in Bio Oss[®] (Figure 1aii), Bone Save[®] (Figure 1bii) and Pro-Osteon[™] 200R (Figure 1cii). A surface coating of fine needle-like crystal conglomerations are visible on Pro-Osteon[™]500R (Figure 1dii) which are typical of highly crystalline HA.

S_{BET} of the bone void filler materials was measured using N₂ gas absorption (Table 2), which takes into consideration only S_{BET} of micro and mesopores. Bone Save[®], Pro-Osteon[™]200R and 500R were relatively low with a range of 2.29 to 9.38 m² g⁻¹, compared to 94.60 and 129.8 m² g⁻¹ for Bio Oss[®] and QUB HA respectively. The S_{BET} values for Pro-Osteon[™] and Bone Save[®] were higher than those (<1 m² g⁻¹) reported in the literature for synthetic sintered HA bioceramics. Bio Oss[®] was found to be in the range of natural bone mineral, reported to be between 87 and 100 m² g⁻¹. QUB HA was also higher than those values (17 - 82 m² g⁻¹) reported in the literature for non-sintered HA [38]. The high S_{BET} of QUB HA can be explained by a smaller particle size and high density of micropores. The bulk porosity of the void filler were characterised using MIP. The results (Table 2) showed all the materials have a porosity of >85%, which was significantly higher than those quoted by the manufacturers (Table 1). Interestingly, the MIP results for Pro-Osteon[™]200R and 500R have given similar porosity values of 87.17 \pm 7.13 and 88.67 \pm 11.84 respectively, despite having

a different pore size range. In this instance, the micro/meso porosity may have played a role in balancing the overall porosities of the two materials.

Chemical analysis: The ICP-MS showed that the main elements in the bone void filler materials were calcium and phosphate with Ca/P molar ratios ranging from 1.48 to 4.44 (Table 3). Human bone mineral has a chemical formula of Ca_{8.3}(PO₄)_{4.3}(OH, CO₃)_{0.3} with a Ca/P molar ratio in the range of 1.5 to 1.70 [39]. The Ca/P and (Ca+Mg)/P molar ratio of Bio Oss[®] and QUB HA was found to be similar to the stoichiometric value of human bone (Table 3). Bio Oss[®] was the only material tested that had the same composition to the mineral phase of human bone. The high molar ratios of 2.90 and 4.44 for Pro-Osteon[™]-200R and 500R are a result of the manufacturing process as only the outer surfaces of the exoskeleton has been converted to HA (Table 1); the bulk of the sample remains CaCO₃, which is evident from the high calcium content found in the ICP results.

Heavy metal contaminants were found to be well below the specified limit of BS ISO 13779-4 and therefore not reported [40]. The presence of magnesium (Mg) ions in apatite structures is important as it plays a key role in bone metabolism; it has a marked inhibitory effect on HA nucleation and growth and stabilizes more acidic precursor phases [41]. Trace levels of Mg were found in all samples. Elevated levels of Mg were evident in Pro-Osteon[™]500R, resulting in a significant increase between Ca/P and (Ca+Mg)/P molar ratios from 4.44 to 5.28 (Table 3). The Ca/P molar ratio of Bone Save[®] corresponds to the theoretical Ca/P ratio of 1.5 for Ca₃(PO₄)₂ materials, confirming the composition given by the manufacturer [32]. The next most abundant metallic ion found in bone is sodium. High trace amounts, ~ 1.6mg/L, of Na ions were found

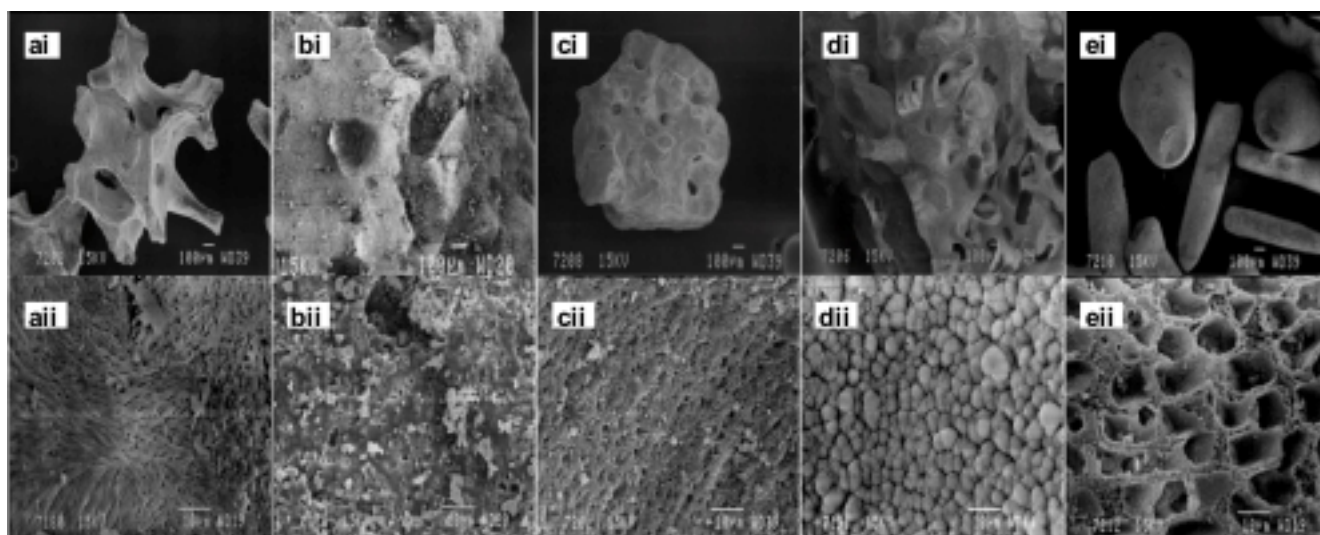


Figure 1: SEM micrographs of (a) Bio Oss (b) Bone Save (c) Pro-Osteon 200R[™] (d) Pro-Osteon 500R[™] and (e) QUB HA; ai-ei: Bars are 100 μ m (Magnification x50); aii-eii: Bars are 10 μ m (Magnification x1500).

Material	Bio Oss [®]	Bone Save [®]	Pro Osteon [™] 200	Pro Osteon [™] 500R	QUB HA
Major Elements					
Ca/P Molar Ratio ^a	1.66	1.48	2.90	4.44	1.61
(Ca+Mg)/P Molar Ratio ^a	1.70	1.49	2.93	5.28	1.70
Minor Elements (mg/L)					
Na	1.90 \pm 0.09	0.91 \pm 0.01	1.61 \pm 0.11	1.98 \pm 0.36	1.70 \pm 0.62
Si	0.22 \pm 0.05	0.56 \pm 0.04	0.22 \pm 0.01	0.25 \pm 0.01	0.40 \pm 0.05
Sr	0.07 \pm 0.02	0.07 \pm 0.01	2.46 \pm 0.87	1.84 \pm 0.08	0.15 \pm 0.02

Table 3: Ca/P Molar Ratio and minor components calculated from element's atomic mass and ICP-MS analysis (n = 6).

in all samples with the exception of the synthetic HA Bone Save®. Other trace elements that were detected at elevated levels were silicon and strontium, both of which are known to enhance bone metabolism [41].

Four of the five bone filler samples investigated using FTIR show vibrational modes corresponding to phosphate, hydroxyl and carbonate groups found in HA (Figure 2). The exception was Bone Save®, which matched those found in synthetic β TCP [31]. In the four HA samples, intense bands at 1010 – 1050 cm^{-1} (strongest in Bio Oss®, QUB HA) and 1426 – 1465 cm^{-1} were assigned to $\nu_2\text{PO}_4^{3-}$, and are consistent with the subgroup of the $\text{P6}_3/\text{m}$ space group of HA [42]. Other characteristic HA bands were observed at 850 – 875 cm^{-1} corresponding to $\nu_2\text{CO}_3^{2-}$ and lower intensity bands at 975 – 990 cm^{-1} shouldering $\nu_2\text{PO}_4^{3-}$ assigned to $\nu_1\text{PO}_4^{3-}$ were also visible [28,42]. The CO_3^{2-} bands at 1465 and 714 cm^{-1} , and 1462 and 713 cm^{-1} in Pro-Osteon™200R and 500R respectively had a much higher intensity compared to the other samples and are likely to belong to unreacted aragonite in the coral [43]. Only one CO_3^{2-} band was observed in QUB HA at 1471 cm^{-1} . In addition a broadband was observed at 3000 - 3500 cm^{-1} which was assigned to O-H stretching and is likely to be a result of water absorption in the sample [44].

XRD diffractograms for the five samples investigated are shown in Figure 3. All samples show the characteristic three peaks residing between 31 and 33° (2 θ), representing (211), (112), (300) planes and a more crystalline peak at 26° (2 θ) representing the (002) plane of HA [45]. The additional peaks were identified as belonging to unreacted calcite or aragonite, matched by the patterns (PDF 47-1473) and (PDF 41-1475) respectively. BioOss® and QUB HA were found to be poorly crystalline, similar to the crystallographic appearance of biological apatites. The sharp peaks found in the other samples indicate the presence of a highly crystalline material, which is a result of the high temperature processing conditions. The additional peaks in the samples were identified as corresponding to magnesium and β TCP phases, matching patterns (PDF 09-0169) and (PDF 47-1473) respectively. Rietveld refinement was applied to the diffractograms to characterize the phase content. The least-square refinement program used for Rietvel showed that all the patterns were well resolved with the calculated profiles. The algal-derived QUB HA was matched to the profile (PDF 73-0294). The phase composition was >95% HA, with minor phases of β TCP and other impurities (Table 3). BioOss® and Bone Save® were found to have phase compositions of HA (100%) and

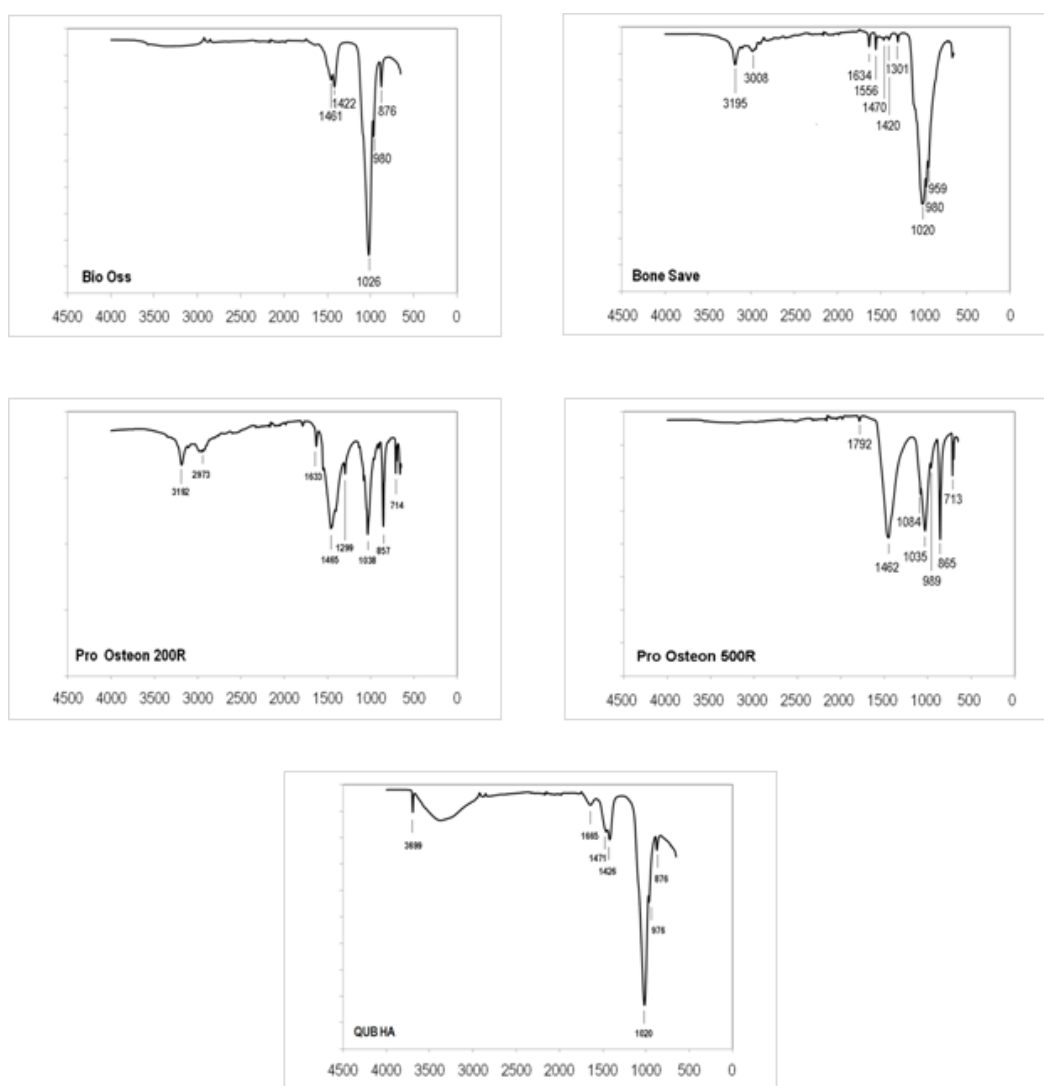


Figure 2: FTIR Spectra of CaP Bone Void Fillers.

β TCP (87.3%); HA (12.7%) respectively. The main phase in the two Pro-OsteonTM samples was aragonite ($82.4 \pm 4.2\%$), with a minor phase of HA, which confirmed the ICP results (Table 3).

The physicochemical results indicated that chemical composition, crystallinity and granule size of QUB HA were most similar to BioOss[®]. These findings support work by Turhani and Weißenböck who also investigated the cell response of a different QUB HA to BioOss[®] [46]. The main differences between QUB HA and Bio-Oss[®] were pore size, granule shape and S_{BET} . All these factors are likely to influence cell response. QUB HA has long uniform-sized cylindrical shaped granules with open microporous extremities, whereas BioOss[®] is irregular in shape with randomly distributed macroporosity. Gonda et al. found that *in vivo*, uniform-sized spherical granules achieved better stability and prognosis than irregular-shaped granules [47].

Biological characterization

Toxicity: Cytotoxicity of the biomaterials was measured after 72

hrs in culture by the release of lactose dehydrogenase (LDH) which occurs upon cell death. No significant difference was found between LDH release in the control cells and those grown on the bone void filler materials (ANOVA, $p = 0.1346$) (Figure 4a). This suggests that the biomaterials are non-toxic to hFOB.

Cell viability: The optical density (O.D.) absorbance values measured from the XTT assay data are given in Figure 4b. A statistically significant increase in hFOB viability with time was observed on the samples with larger granule sizes, namely Bone Save[®] ($p = 0.008$) and Pro-OsteonTM500RTM ($p = 0.014$). No significant difference was observed with time on the BioOss[®]. After 7 days in culture, hFOB viability was significantly higher in Pro-OsteonTM500RTM and QUB HA compared to the other biomaterials ($p < 0.001$). A slight increase in cell viability in Pro-OsteonTM500RTM compared to the cells only control is likely to be caused by the larger surface area of the 3D granules compared to a 2D flat surface although this difference was not significant ($p = 0.078$). Our results indicate that the microporous algal-HA was effective in

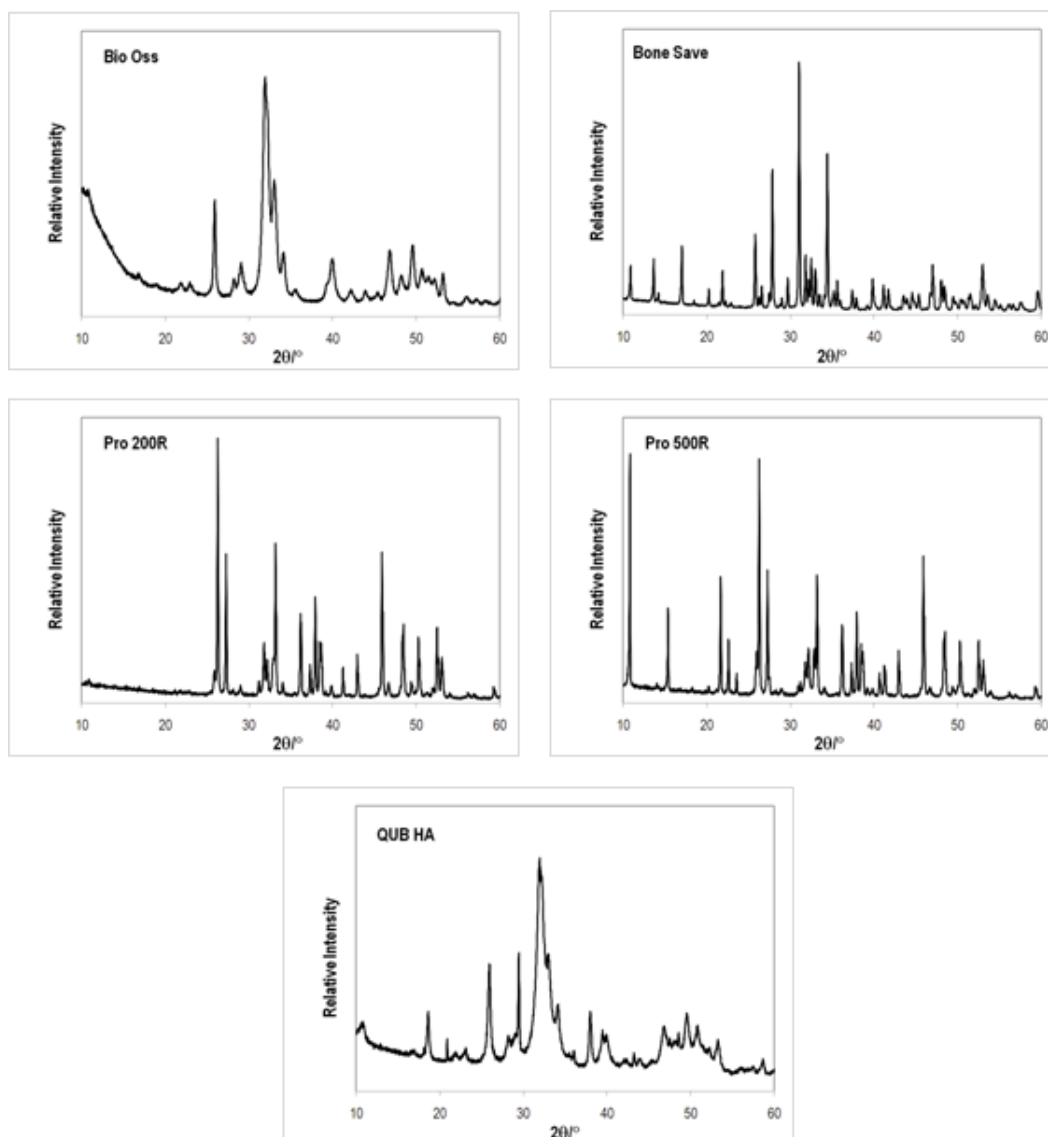


Figure 3: XRD traces of CaP Bone Fillers.

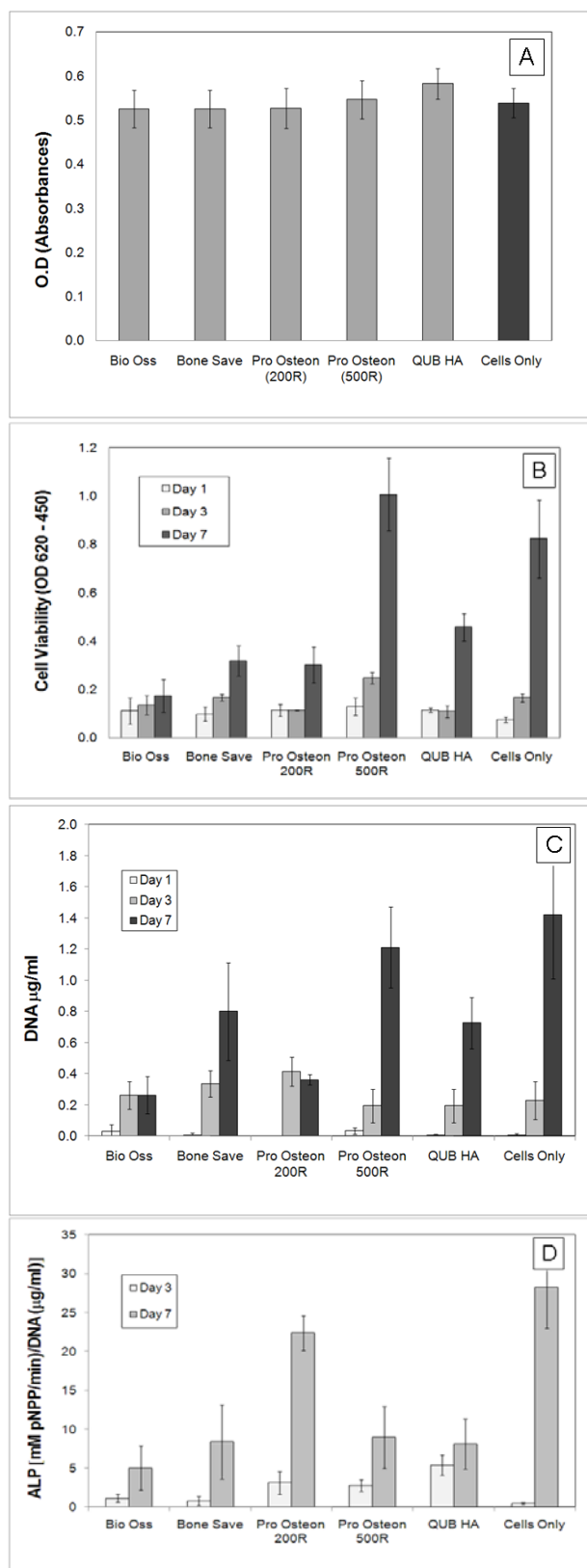


Figure 4: Biological Evaluation (a) LDH Cytotoxicity Assay (b) Optical density measurements for XTT cell viability assay as a function of time (c) Cell proliferation assay incubated at 33°C and (d) ALP Activity normalized to DNA (Picogreen Assay) at 39°C.

supporting cell viability as that of its macroporous counterparts. This contradicts findings by Isaacset al. [13] who reported that microporosity in β TCP negatively modulates cell viability. This would suggest that other factors such as S_{BET} , particle size and chemical composition also heavily influence the cellular response. For example, the algal-HA we tested was found to have a significantly higher S_{BET} of $129.8 \text{ m}^2 \text{ g}^{-1}$, than that of those tested by Isaacs et al. who reported their materials to have S_{BET} of 0.001 to $1.73 \text{ m}^2 \text{ g}^{-1}$.

Cell proliferation: The proliferative ability of hFOB cells on the samples was determined by picogreen assay (Figure 4c). At day 1, the cell activity was low in all groups. Between day 1 and day 7, the number of cells in all groups increased significantly, indicating that the biomaterials support proliferation. hFOBs showed the best proliferative capacity on Pro-Osteon 500RTM in comparison to the other bone void filler materials. At day 3, no significant difference was observed between any of the materials or cell controls (ANOVA, $p = 0.101$). At day 7, no significant difference was found between Bone Save[®] and Pro-OsteonTM500RTM, and the cells-only control ($p = 0.145$, $p = 0.127$). However, the proliferative capacity of hFOBs on QUB HA increased significantly compared to Bio-Oss[®] and Pro-OsteonTM200R. When comparing the smaller granules at 7 days, QUB HA showed a better proliferative capacity. Interestingly, proliferation was significantly lower on Pro-Osteon 200RTM compared to Pro-OsteonTM500RTM at day 7, despite having a similar chemical composition (Table 2). This indicates that particle size has a significant influence on cell behaviour *in vitro* and supports work by Weiß enböck et al., who found that particle size and substrate geometry significantly influenced cell signaling [30].

Cell differentiation: ALP is one of the most common markers of osteogenic differentiation. ALP activity is expressed as ALP/mg DNA giving the ALP activity per cell, to adjust for significant differences in cell numbers between groups (Figure 4d). A significantly higher level of ALP activity was found between day 3 and 7, with the exception of QUB HA indicating that the cells may differentiate earlier on QUB HA. No significant differences were found in the level of ALP activity between BioOss[®], Bone Save[®], Pro-OsteonTM 500R and QUB HA at day 7 ($p = 0.397$). Pro-OsteonTM200R showed a significant increase in ALP activity when compared to these groups, but no significant difference when compared to the cells-only control ($p = 0.131$). The physicochemical results showed that QUB HA was most similar to Bio-Oss[®], however the cell analysis showed that QUB HA supported more cell growth at 7 days and more ALP at day 3.

ALP activity was low on all materials when compared to the cells-only controls after 7 days with the exception of Pro-OsteonTM200R. These results, in conjunction with those for cell proliferation, suggest that hFOBs on Pro-OsteonTM200R differentiate earlier than those on the other materials. As cells differentiate, their proliferative capacity decreases, therefore the delayed differentiation seen on all other materials could contribute to the increase in cell number found on Pro-OsteonTM500R, Bone Save and QUB HA.

In vitro results: The results showed that cells maintain their function on the microporous QUB HA with similar or better biocompatibility than the clinically available bone void fillers tested. In our *in vitro* studies, instability of bone void fillers could have caused disruption to the cell monolayer formation, thus affecting cell attachment and proliferation but the rationale of testing the fillers without a fixative agent (e.g. fibrin glue) was to ensure that the cell response was a direct result of the filler. The use of fixatives also had the potential to block pores, thus reducing the topographical effect of the fillers.

Some studies have suggested that larger microporous surface areas increase protein adsorption, enhancing cell attachment [48], however others have argued that protein adsorption does not enhance cell attachment but promotes inducible cells to differentiate down the osteogenic lineage [10]. The S_{BET} was significantly higher, $129.8 \text{ m}^2 \text{ g}^{-1}$ for QUB HA, compared to $94.60 \text{ m}^2 \text{ g}^{-1}$ for Bio Oss® and we found that there was an increase in proliferation on QUB HA but similar levels of ALP expression, suggesting the former effect is more likely.

Our results also suggest that granule size has a significant effect on the cellular response. By comparing Pro-Osteon™ 200R to 500R™ it is possible to assess the influence of different granule sizes on cellular behaviour. The physicochemical data showed that Pro-Osteon™ 200R and 500R™ have the same pore size, surface area and chemical composition but differ in size of granule (Table 2) however; the biochemical results have shown a significantly higher proliferation of cells on Pro-Osteon™ 500R compared to 200R, whereas differentiation was significantly lower in Pro-Osteon™ 500R compared to Pro-Osteon™ 200R. This may be directly related to the granule size but it is important to note that, although Rietveld analysis quantified the materials with the same phase composition, ICP analysis (Table 3) found elevated levels of Mg in Pro-Osteon™ 500R. Other studies have shown that Mg enhances osteoconductivity by stimulating better cell adhesion and proliferation, and could also account for the increased cell number [20]. All materials tested were HA, with the exception of Bone Save® which was predominantly β TCP with a minor HA phase. This, and other studies, have shown that the manufacturing process used to fabricate bioceramics strongly influences chemistry and pore geometries and thus the cellular response.

Conclusion

This was a pilot study to assess a novel sustainable microporous algal HA bioceramic as potential bone void filler. Outcome measures were the ability to support proliferation and differentiation of hFOB cells. It was hypothesized that microporous granules would perform equally to macroporosity in terms of the biological response. Although direct comparison between micro and macro scale porosity was difficult due to other chemical and physical variation between materials, the results have clearly shown that the algal-HA (QUB HA) is capable of supporting cellular activity. QUB HA exhibited a similar rate of osteogenic differentiation to all other biomaterials tested, with the exception of Pro-Osteon™ 200R. It also showed an increased rate of proliferation and cell viability compared to the other biomaterials, with the exception of Pro-Osteon™ 500R. One possible explanation is that the interstitial macroporous space between QUB HA granules has provided sufficient interstitial space for cells to penetrate between the granules and fulfil their functions, however this theory needs to be tested *in vivo*. Our results have clearly shown that HA granules with an interconnected microporous morphology, which provides a high surface area, positively supports cell viability, proliferation and differentiation. Furthermore the results suggest that other factors, such as granule size, specific surface area and surface chemistry, are equally as important as pore size in tissue regeneration. The next stage of this work will focus on *in vitro* bioresorption profiling of these materials followed by *in vivo* trials.

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